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ABSTRACT

Single-crystal PbTiO₃ layers were grown on (001) SrTiO₃ substrates by molecular beam epitaxy using hydrogen peroxide as an oxidant. Phase composition and structural properties of the films were examined as a function of growth parameters. It was found that single-phase PbTiO₃ films grew epitaxially at substrate temperatures of 600°C and higher, whereas layers grown at lower temperature contained PbO inclusions. The epitaxial relationship between the films and the substrates was (001)PbTiO₃//(001)SrTiO₃ and PbTiO₃[100]//SrTiO₃[100]. No evidence of a-domains was found. Full widths at half maximum of (001) PbTiO₃ rocking curves were as low as 6-8 arcmin for 50-60 nm thick films, indicating high crystal quality of the films. From ellipsometric measurements, refractive index and band gap of PbTiO₃ were found to be 2.66 at 633 nm and 3.8 eV, respectively.

Growth of Pb(Zr_xTi_{1-x})O₃ (PZT) films by molecular beam epitaxy was demonstrated for the first time. Single-crystal, single-phase PZT films were grown on (001) SrTiO₃ substrates at a growth temperature of 600°C. *In situ* monitoring of the growth process by reflection high-energy electron diffraction revealed two dimensional growth mode for the end compounds of the PZT system, PbTiO₃ and PbZrO₃, and three-dimensional growth mode for PZT films of intermediate compositions. Layer-by-layer growth of PZT films was achieved by using a PbTiO₃ buffer layer between the SrTiO₃ substrate and PZT films. Optical properties of the films of end compositions were investigated by spectroscopic ellipsometry. Refractive index at 633 nm was found to be 2.66 for PbTiO₃ and 2.40 for PbZrO₃. Band gap energies of PbTiO₃ and PbZrO₃ are 3.81 and 3.86 eV, in good agreement with theoretically calculated values. The P-E hysteresis loop of the 70-nm-thick PZT film was well saturated and had a square shape. The remanent polarization and the coercive field were 83 μ C/cm² and 77 kV/cm, respectively.

Epitaxial growth of PbO, TiO₂ and ZrO₂ has been achieved on MOCVD grown GaN template using oxides MBE with a reactive H₂O₂ oxygen source. In situ RHEED was used to monitor the growth in-situ. AFM was used to characterize the surface morphology of the thin PbO and ZrO₂, which show streaky, 2-D RHEED patterns. XRD pattern indicates that the growth orientation of these oxides are PbO [111]//GaN [0002], ZrO₂[100]//GaN [0002] and TiO₂[200]//GaN[0002].

As the abstract indicates, we have made substantial progress in the growth and characterization of PTO and PZT and also on ZrO₂ which turned out to be a very good oxide bridge layer to GaN. Below, the details of PTO, then PZT followed by ZrO₂ are discussed.

PTO

Lead titanate (PbTiO₃), a ferroelectric material with perovskite structure, has received a great deal of attention owing to a unique combination of its piezoelectric, pyroelectric, dielectric, electo- and acousto-optic properties [¹]. PbTiO₃ is a very attractive material for use in a wide variety of fields, including ultrasonic sensors, infrared detectors, electro-optic modulators, and ferroelectric random access memories. To harness the intrinsic properties of PbTiO₃ for device applications, however, high-quality epitaxial films are required. PbTiO₃ thin films have been prepared by various methods such as metal organic chemical vapor deposition (MOCVD) [²], rf magnetron sputtering [³], pulsed laser deposition [⁴], hydrothermal method [⁵], and sol-gel technique [⁶]. Surprisingly, molecular beam epitaxy (MBE), a modern growth technique providing high crystal perfection and precise control over material composition, has not been widely applied, except of a few reports involving ozone as an oxidizing agent [^{7 8}]. In this contribution, we report growth of high-quality single-crystal PbTiO₃ layers on (001) SrTiO₃ substrates by MBE utilizing hydrogen peroxide as an oxidant, initially developed for ZnO [⁹].

Experimental details for PTO

PbTiO₃ layers were grown on (001) SrTiO₃ substrates in a Riber 3200 MBE system. The chamber was evacuated by a 1200 l/s turbo pump to a base pressure of 10⁻⁸ Torr. Lead (6N purity) and titanium (6N purity) were supplied from double-zone and high-temperature effusion cells, respectively. A 50% aqueous solution of hydrogen peroxide was used as a source of reactive oxygen [9]. A jet of H₂O₂/H₂O vapor was directed at the substrate through a quartz pipe connected to a leak valve.

Before loading into the chamber, single-crystal SrTiO₃ substrates were etched with a buffered NH₄F-HF solution, rinsed in deionized water and dried with nitrogen. Then, the substrates were loaded into the growth chamber and heated at 600° C for 20 min under H₂O₂/H₂O pressure of 1×10^{-5} Torr. The PTO layers were grown at an H₂O₂/H₂O pressure of about 5 x 10^{-5} Torr, and a Pb-to-Ti flux ratio >>1. The substrate temperature, T_S, was varied in the range from 580 to 620°C and was measured by a thermocouple located at the back of the substrate. The growth process was monitored *in situ* by reflection high-energy electron diffraction (RHEED). The thickness of the grown film varied from 24 to 65 nm.

The crystal structure of the grown layers was examined by high-resolution x-ray diffractometry (XRD) (CuK_{α} radiation). The surface morphology of the samples was

studied by atomic force microscopy (AFM). The thickness of the grown layers was measured with an Alpha-step 250 profilometer. Spectroscopic ellipsometric measurements were performed with variable angle spectroscopic ellipsometry (VASE) at room temperature and at incidence angles of 65, 70, and 75°.

Results and discussions for PTO

Figure 1 shows the typical RHEED patterns of a SrTiO₃ substrate and a PbTiO₃ layer grown at 600 and 620 °C along the [110] and [100] azimuths. The streaky patterns in Figures 1 (c) and (d) indicate that the growth proceeds in a two-dimensional mode. A two-fold surface reconstruction was observed along the [110] direction (Figure 1 (d)). Surface roughness of the PbTiO₃ films measured by AFM varied from 1.3 to 2 nm.

Figure 2 illustrates the effect of the growth temperature and metal flux on the film composition. The XRD data indicate that the film grown at 580° C consists of a mixture of PbTiO₃ and PbO phases (Fig. 2 (a)). The undesirable PbO can be removed by *ex situ* post-growth annealing of the films. It should be mentioned, however, that the post-growth annealing leads also to pronounced surface roughening, which deteriorates the quality of the layers. The films grown at higher temperatures are single-phase and *c*-axis oriented. Only (001) reflections of PbTiO₃ and of SrTiO₃ are visible in the θ -2 θ

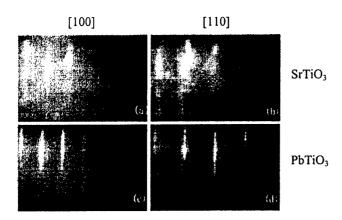


Figure 1. Representative RHEED patterns of (a) and (b) an SrTiO₃ substrate and (c) and (d) a PbTiO₃ layer, respectively.

scans for the films grown at 600° C (Figure 2 (b)). However, (h00) reflections of PbTiO₃ should coincide with (00l) reflections from the SrTiO₃ substrate. So, to detect a-domains possibly present in the films, rocking curves of SrTiO₃ (001) reflections were measured. No evidence of a-domains was found, which can be explained by the fact that all our films were thinner than the critical thickness for the 90° domain formation (100-150 nm [2,10]).

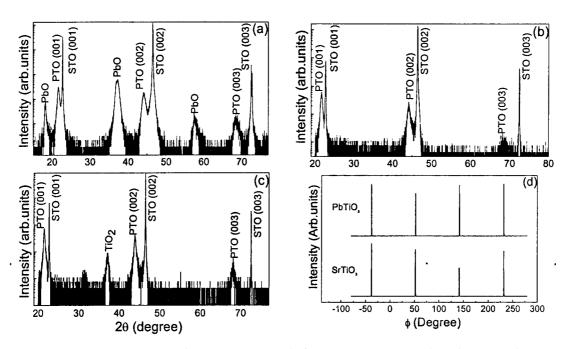


Figure 2. XRD θ-2θ scans for PbTiO₃ (PTO) films grown on SrTiO₃ (STO) substrates at temperature of (a) 580°C, (b) 600°C, (c) at 600°C and increased metal fluxes, peroxide pressure being the same; and (d) Phi-scan for (101) planes of SrTiO₃ substrate and PbTiO₃ layer.

Figure 2 (c) shows the θ -2 θ scan for a film grown at a substrate temperature of 600°C, but with increased Ti and Pb fluxes (Pb/Ti flux ratio and H₂O₂/H₂O pressure were the same as for the samples shown in Figures (a) and (b)). One can see that the film grown at higher metal flux contains the TiO₂ phase in addition to PbTiO₃. This result is similar to that obtained by Theis et al. [7,8], who have reported that, despite excess Pb supply, TiO₂ phase is formed under low ozone pressures due to insufficient Pb oxidation. Taking into account that oxidation of Ti is more favorable because of the higher binding energy of Ti-O bonds [11], the appearance of TiO₂ can be attributed to the fact that available oxidant pressure was insufficient to oxidize increased amount of Pb. However, our finding indicates that the real situation is probably even more complicated. When a PbO film grows at the same Pb flux, substrate temperature, and H₂O₂/H₂O pressure, the same amount of peroxide is sufficient to provide efficient oxidation of much larger amounts of Pb. Therefore, the growth rate of PbO is five times larger than that of PbTiO₃ (60 nm/h for PbO as compared to 12 nm/h for PbTiO₃). The phase composition of the PbO films was verified by XRD. To clarify this issue, we evaluated the probability of incorporation of Pb atoms into PbTiO₃ and PbO. Incorporation coefficient, I, was calculated as

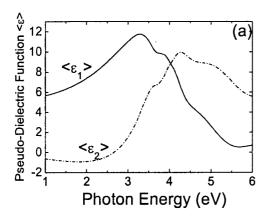
$$I = J_{arr}/J_{inc}, (1)$$

where J_{arr} and J_{inc} are fluxes of Pb atoms arrived to the surface and incorporated into a growing film, respectively. J_{arr} is equal to the product of Pb deposition rate and Pb atomic density (3.28 x 10^{22} cm⁻³), and J_{inc} is equal to the product of PbTiO₃ or PbO growth rate and the density of Pb sites in PbTiO₃ or PbO lattice. The coefficients of Pb incorporation

into PbTiO₃ and PbO were estimated to be 0.017 and 0.136, respectively. One can see that the incorporation of Pb into PbTiO₃ is lower than that into PbO by a factor of 8, indicating complicated nature of the growth mechanism of PbTiO₃. We can presume that such factors as density and configuration of dangling bonds on the surface, surface reconstruction, presence of Ti, and growth mode (three-dimensional growth of PbO versus layer-by-layer mode for PbTiO₃) can influence the incorporation of Pb. Much remains to be learned about the growth mechanism.

Figure 2 (d) shows Phi scans for (101) reflections of a SrTiO₃ substrate and a PbTiO₃ film grown at 600°C. One can see that both the layer and the substrate show four peaks 90° apart from one another. The peaks from the film and the substrate are at the same angular positions, indicating the epitaxial relationship (001)PbTiO₃//(001)SrTiO₃ and PbTiO₃[100]//SrTiO₃[100]. Lattice parameters c and a were calculated from 20 positions of symmetrical (001) and asymmetrical (101) reflections, respectively. For our PbTiO₃ films in the thickness range of 24 - 65 nm, c-parameters were found to vary from 4.10 to 4.13 Å, and a-parameters from 3.91 to 3.93 Å. The values of c for all our films are less that the reported bulk c-parameter (c = 4.156 Å [12]), whereas a-parameters are larger than the bulk value (a = 3.905 Å [12]), indicating that the films are under in-plain tensile strain. Since the lattice mismatch between PbTiO₃ and SrTiO₃ is very small, the residual strain arise most probably during sample cooling after growth due to the difference in thermal expansion coefficient between the film and the substrate or/and to the volume change during transformation from the cubic to tetragonal PbTiO₃ phase. Can we be more quantitative by figuring out the strain due to cooling and or phase change. Full width at half maximum (FWHM) of XRD rocking curves measured for the (001) reflection decreased from 16 arcmin to 6 arcmin as the film thickness increased from 24 to 65 nm, pointing to a very good alignment in the growth direction. The value of 6 arcmin is comparable to the best values reported for MOCVD [2] and hydrothermal [5] PbTiO₃ films.

Pseudodielectric function of PbTiO₃ was measured by spectroscopic ellipsometry in the spectral range form 1 to 6 eV. The sample structure was modeled as consisting of a rough surface layer, a main PbTiO₃ layer, and a substrate, and the dielectric function of the main PbTiO₃ layer was estimated using the parametric optical constant (POC) model [¹³]. The pseudodielectric functions and modeled dielectric functions match very well and are not discernible because of the excellent fitting, as seen from Figure 3, which shows the pseudodielectric function and fitted dielectric function for a PbTiO₃ film grown on a SrTiO₃ substrate.



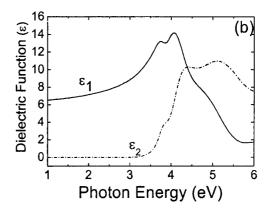
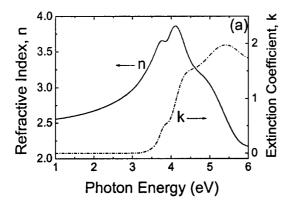


Figure 3. (a) Pseudodielectric function ($\langle \epsilon \rangle = \langle \epsilon_1 \rangle + i \langle \epsilon_2 \rangle$) of the PbTiO₃ film grown on (100) PbTiO₃ substrate. Solid and dash-and-dotted curves represent the real and imaginary parts, respectively. (b) Fitted dielectric function ($\epsilon = \epsilon_1 + i\epsilon_2$) of the PbTiO₃.

The thickness of this layer was fitted to be 35.5 ± 0.9 nm, and the surface roughness layer, 4.0 ± 0.3 nm. Figure 4 (a) presents fitted complex refractive index of the 35-nm-thick PbTiO₃ layer. Note that the dielectric function is the square of the complex refractive index $\varepsilon = (n+ik)^2$. The value of the refractive index is 2.66 at 633 nm (1.96 eV), which is very close to the bulk value of 2.668 [¹⁴], indicating excellent crystal quality of the material.



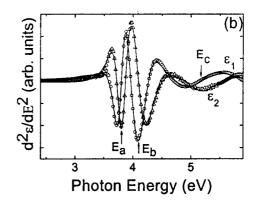


Figure 4. (a) Fitted complex refractive index (n+ik) of PTO layer. (b) Plot of the second derivative of the PbTiO₃ dielectric function (symbols) and their fits (sold curve) of using SCP model. The fitted band-gap energies (E_a , E_b , and E_c) are marked with arrows.

In order to estimate the band-gap parameters of PbTiO₃, we took the second derivative of the dielectric functions and fitted them using the standard critical point (SCP) model [¹⁵]. The SCP model assumes simple parabolic dispersion relation for direct interband transitions and provides accurate critical point parameters (band gap energies). Figure 4 (b) presents the second derivative of the dielectric functions and their fit using the SCP model. The fitted band-gap energies (E_a, E_b, and E_c) are marked with arrows and listed in Table I, along with the theoretical values found previously by Lee *et al.* [¹⁶], who

calculated the band structure of PbTiO₃ by using local density approximation (LDA) method. According to their calculation, there are four interband transitions in PbTiO₃ in the visible-ultra-violet spectral range. They estimated the transition amplitudes and concluded that only three transitions are substantial. As shown in the Table I, the ellipsometrically estimated band gap values are in good agreement with those calculated theoretically.

Table I. Experimental and theoretical PbTiO₃band gap parameters.

Band-gap energy	Transition	Experimental	Theory [16]
		(this work)	
Ea	$X_{5'\nu} \to X_{3c}$	$3.81 \pm 0.01 \text{ eV}$	3.78 eV
E _b	$X_{4'\nu} \to X_{1c}$	$4.11 \pm 0.01 \text{ eV}$	4.00 eV
E_c	$X_{5'v} \to X_{1c}$	$5.2 \pm 0.1 \text{ eV}$	4.93 eV

Conclusions for PTO

High-quality PbTiO₃ films were grown on SrTiO₃ substrates by MBE utilizing H₂O₂ as an oxidant. Conditions for the MBE growth of single-phase, single crystalline PbTiO₃ films were found: the films grown at 580°C contained lead oxide inclusions, whereas single-phase layers grew at 600 and 620°C with the epitaxial relationship (001)PbTiO₃//(001)SrTiO₃ and PbTiO₃[100]//SrTiO₃[100]. The low values of FWHM of (001) rocking curves (down to 6 arc min) are indicative of high crystal quality of the grown layers. The lattice constants of PbTiO₃ determined from XRD data indicate the films to be under in-plane tensile strain. Refractive index and band gap energies of the PbTiO₃ films were determined with variable angle spectroscopic ellipsometry in the spectral range of 1 to 6 eV. Refractive index was found to be 2.66 at 633 nm, which is close to the bulk value. The experimentally determined band gap values agree well with the band structure calculations performed previously.

PZT

Introduction for PZT

Due to their large piezoelectric coefficient, electrical polarization, and electromechanical coupling factor, ferroelectric Pb(Zr_xTi_{1-x})O₃ (or PZT) thin films are of considerable interest for a wide range of applications, such as gate material for ultrasonic sensors, infrared detectors, ferroelectric filed effect transistors, and nonvolatile ferroelectric random access memory devices [¹⁷]. Piezoelectric and ferroelectric properties of this material have been studied extensively as a function of composition. It was demonstrated that the piezoelectric coefficient, relative permittivity, and remanent polarization peaked

near the morphotropic phase boundary [¹⁸]. To exploit the unique properties of PZT for device applications, high quality single-crystal PZT films are required. However, most of the studies have been done on ceramic samples, whereas the properties of single-crystal thin films can differ drastically from those for ceramics [¹⁹] and depend strongly on film orientation [²⁰, ²¹, ²², ²³].

PZT thin films have been prepared by various methods such as metal organic chemical vapor deposition (MOCVD) [20,21, 22], rf magnetron sputtering [²⁴, ²⁵, ²⁶], pulsed laser deposition [²⁷, ²⁸], and sol-gel technique [²⁹, ³⁰]. To the best of our knowledge, molecular beam epitaxy (MBE), a modern growth technique providing high crystal perfection and precise control over material composition, has not been used for the fabrication of PZT films up to now. In this contribution, we report on the growth of high quality PZT films by peroxide MBE and study of their structural, optical, and ferroelectric characteristics.

Experimental details for PZT

PZT layers were grown on (001) SrTiO₃ substrates in Riber 3200 MBE system. The chamber was evacuated by a 1200 l/s turbo pump to a base pressure of 10⁻⁸ Torr. Aqueous solution of hydrogen peroxide (H₂O₂) was employed as a source of reactive oxygen [31], 6N-purity lead and titanium were supplied from double-zone and hightemperature effusion cells, respectively. Due to the very low equilibrium pressure of metal Zr, zirconium tetra (Zr-t) butoxide was chosen as the metal-organic precursor, and 6N-purity Ar was used as a carrier gas. Before loading into the chamber, SrTiO₃ substrates were etched with a buffered NH₄F-HF solution, rinsed in deionized water and dried with nitrogen. Then, the substrates were loaded into the growth chamber and heated at 600°C for 20 min under H₂O₂/H₂O pressure of 1x10⁻⁵ Torr. PZT layers 30 to 70 nm thick were grown at an H₂O₂/H₂O pressure of about 5 x 10⁻⁵ Torr, a substrate temperature of 600°C, and a Pb-to-Ti flux ratio >>1. Zr/Ti ratio in the PZT films was varied by changing a flow rate of the carrier gas. The growth process was monitored in situ by reflection high-energy electron diffraction (RHEED). Surface morphology of the films was examined by atomic force microscopy (AFM). Phase composition and structural properties of the films were examined by x-ray diffraction (XRD). The thickness of the grown layers was measured with an Alpha-step 250 profilometer. Spectroscopic ellipsometric measurements were performed with the use of variable angle spectroscopic ellipsometry (VASE) at room temperature and incidence angles of 65, 70, and 75°.

Results and discussions for PZT

Figure 1 shows RHEED patterns of the end compounds of the Pb(Zr_xTi_{1-x})O₃ system, PbTiO₃ and PbZrO₃, taken after growth. The streaky RHEED patterns indicate that the growth of these compounds proceeded in two-dimensional mode. However, when PZT films were grown directly on SrTiO₃, spotty RHEED patterns were observed in the beginning of the growth due to three-dimensional growth mode and rough surface morphology (Figure 2 (b)). The spotty RHEED patterns of the PZT films became worse with time, transforming into partially ring pattern characteristic of textured polycrystalline films, as illustrated in Figure 2 (c). To overcome this problem, we introduced PbTiO₃ buffer layers between the SrTiO₃ substrates and PZT films. As seen

from Figure 3, the PZT layer grown on the PbTiO₃/SrTiO₃ structure show streaky RHEED patterns characteristic of two-dimensional growth and smooth film surface. Surface morphology of the PZT films grown through the buffer layers was examined by AFM. Figure 3 (d) shows 1x1 μm AFM image of the PZT film surface. Steps and terraces are visible on the surface. The step height is about 4 nm, which is comparable to the unit cell of PZT, evidencing layer-by-layer growth mode. The root mean square of surface

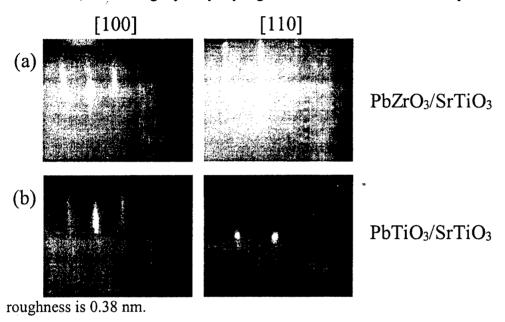


Figure 1. Representative RHEED patterns of (a) PbTiO₃ and (b) PbZrO₃ films grown on SrTiO₃ substrates.

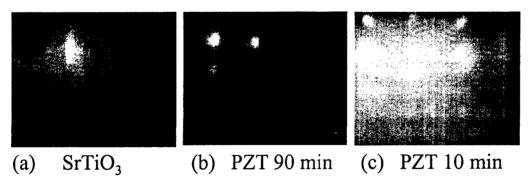
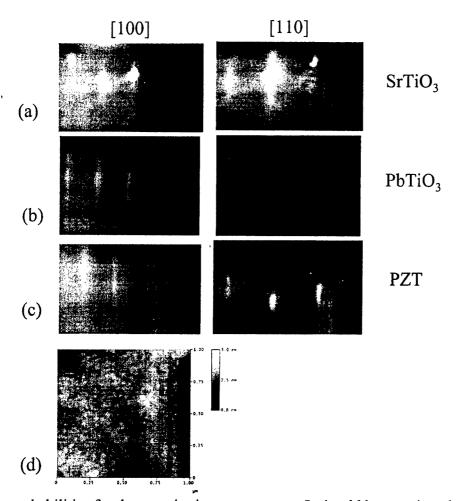


Figure 2. RHEED patterns taken along [100] azimuth from (a) SrTiO₃ substrate, (b) PZT layer after 10 min growth, and (c) PZT layer after 90 min growth.

As evidenced from XRD data, the PZT films of all investigated compositions are single-phase, c-axis oriented. Figure 4 presents θ-2θ scans of a PZT/PbTiO₃/SrTiO₃ and PbZrO₃/SrTiO₃ structures. Only (00*l*) reflections of the substrate and the layers are seen in the scans. As mentioned in the Experimental section, the Zr content in the PZT films was varied by changing flow rate of the carrier gas and was roughly estimated from

growth rates of end compounds, PbTiO3 and PbZrO3, assuming the same incorporation



probabilities for the constituting components. It should be mentioned, that the estimation of PZT film composition from XRD data is a severe problem, because our films are very thin (below 100 nm) and not fully relaxed, so thickness and residual strain resulted from lattice mismatch, difference in thermal expansion coefficients, and volume change during phase transformation have significant influence on the lattice parameters. Therefore, another technique, such us x-ray fluorescence spectrometry, secondary ion

Figure 3. RHEED patterns of (a) SrTiO₃ substrate, (b) PbTiO₃ buffer layer, and (c) a 30-nm-thick PZT layer taken after growth, and (d) AFM image of this PZT film, vertical scale is $5 \mu m$.

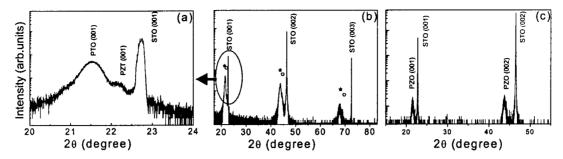


Figure 4. XRD θ-2θ scans for a PZT film grown on a PbTiO₃ (PTO) layer on SrTiO₃ (STO) substrates: (a) first-order reflections and (b) wider angular range for the same scan (PbTiO₃ and PZT reflections are marked by an asterisk and a circle, respectively); and (c) for a PbZrO₃ layer on SrTiO₃ substrate.

mass spectrometry, or Rutherford backscattering, will be applied to determine exact film compositions.

Optical properties of PbTiO₃ and PbZrO₃ films were investigated by spectroscopic ellipsometry. The ellipsometric measurements as well as results for PbTiO₃ films are described in more detail in Ref [³²]. Refractive index at 633 nm is found to be 2.66 for PbTiO₃ and 2.40 for PbZrO₃, which is consistent with the literature data [³³]. The experimentally determined band gap values 3.81 eV for PbTiO₃ and 3.86 eV for PbZrO₃ are in good agreement with theoretical values calculated previously by Lee *et al.* [³⁴]. Optical properties of the films of intermediate PZT compositions are now under examination.

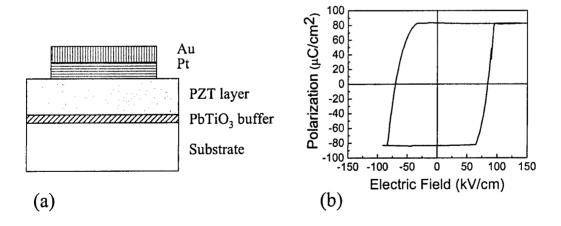


Figure 5. (a) Schematic sketch of the capacitor test structure. (b) P-E hysteresis curve for the 70-nm-thick PZT film.

To investigate ferroelectric properties, PZT films were grown on conductive Nb-doped $SrTiO_3$ substrates. Capacitor device structures were fabricated as shown in Figure 5 (a). Au/Pt top electrodes 30nm/30nm in thickness, 300 μ m in diameter were deposited by ebeam evaporation. The polarization *versus* applied electric field (P-E) characteristics

were measured at room temperature with a Radiant Technologies Precision LC ferroelectric test system. Figure 5 (b) shows a P-E hysteresis loop for the 70-nm-thick PZT film. The loop is well-saturated and square in shape, with high remanent polarization of 83 μ C/cm², and the coercive field of 77 kV/cm. It should be mentioned, however, that current-voltage characteristics of this structure shows high leakage current of 5.7×10^{-2} A/cm² for applied bias of 2 V. The reason of the high leakage is now under investigation.

Conclusions for PZT

Single crystal, single-phase PZT films were successfully grown on (001) SrTiO₃ substrates by peroxide MBE. It is found that the end compounds of the Pb(Zr_xTi_{1-x})O₃ system, PbTiO₃ and PbZrO₃, grow on SrTiO₃ in two dimensional mode, whereas the growth of PZT films of intermediate compositions proceeds in three-dimensional mode. The use of PbTiO₃ buffer layer between the SrTiO₃ substrate and PZT results in layer-by-layer growth of PZT films. From ellipsometric measurements, values of refractive index and band gap energy were found for the end compositions of the PZT system. Square-shaped P-E hysteresis loop was observed for the 70-nm-thick PZT film, with a remanent polarization of 83 μ C/cm². The large value of the remanent polarization evidences that the MBE-grown PZT films show a great promise for device applications.

ZrO₂

Introduction for ZrO₂

Lead titanate based thin films, such as PbTiO₃ (PTO) and Pb (Zr,Ti)O₃ (PZT) show a great promise for a wide variety of applications such as integrated piezoelectric devices³⁵, pyroelectric infrared sensors³⁶, nonvolatile memory³⁷, optical waveguide devices,³⁸ and electro-optic modulators.³⁹ The integration of these perovskite oxides with existing semiconductors such as GaN is of particular interest which makes use of intrinsic characteristics of ferroelectric thin films, in the form, *e.g.*, PZT gated GaN based ferroelectric MODFET⁴⁰.

Experimental details for ZrO₂

Due to the very different crystal structure between the PZT and GaN (perovskite for PZT while hexagonal for GaN), a bridge layer must be inserted for their successful integration. PbO, ZrO₂ and TiO₂ are interesting materials by themselves and each of them has their applications in different fields. In the presented work, these oxides are of particular interest due to the factor that they are relative compounds of PZT and therefore could be promising candidates for the integration of PZT and GaN, as they have been used in the literatures as the buffer layer for PZT growth⁴¹, ⁴², ⁴³. We present a study of PbO, TiO₂ and ZrO₂ grown by molecular beam epitaxy (MBE) as candidates for bridge layers for the integration of PZT with GaN. The GaN template used in this experiment was prepared by

metal-organic chemical vapor deposition. The MBE system used in this study is a Riber 3200 customized for oxides growth. Ti flux was provided by a high-temperature Knudsen cell and Pb was evaporated from a double-zone Knudsen cell. Metal-organic Zr precursor, Zr-t butoxide, was kept in a bubbler heated by a water bath and argon was used as the carrier gas. In order to overcome the shortcomings of the conventional oxygen plasma sources such as possible surface damage by high-energy particle and degradation of plasma sources due to the high reactivity of oxygen radicals, a highly reactive H_2O_2 source was employed. 50% aqueous solution of H_2O_2 was placed in a stainless steel bubbler heated by water bath. The flow rate of the H_2O_2 was controlled by the leak valve and the pressure was monitored by the vacuum gauge. A schematic of our growth chamber is shown in figure 1.

Results and discussions for ZrO₂

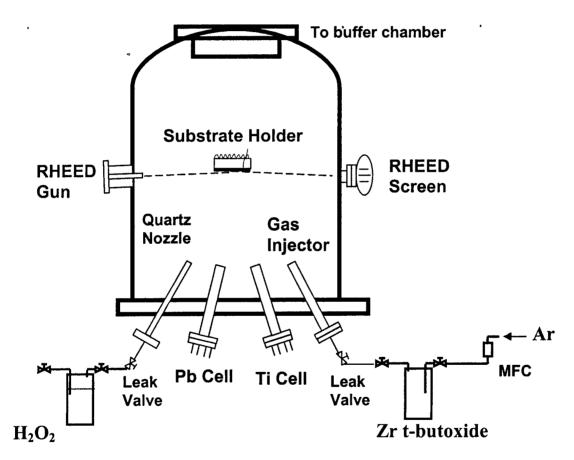
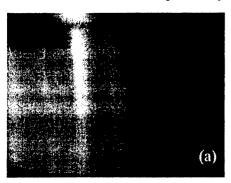


Figure.1 Schematic drawing of our Riber 32 system for oxides growth

Different approaches were used for epitaxial growth of PbO, TiO₂, and ZrO₂ was performed using: epitaxy of TiO₂ was performed at 600°C. A two step growth was found to be needed for ZrO₂: growth of low temperature buffer at 300°C to minimize the oxidation of GaN template followed by annealing at 750°C and high temperature growth at 700°C. For both TiO₂ and ZrO₂ growth, the H₂O₂ pressure was kept between 1.5 and 3.0 x 10⁻⁵ Torr. The situation is different for the growth of PbO, which is more volatile

than both TiO2 and ZrO₂. Therefore, a reduced temperature (550°C) and high H_2O_2 pressure (~5.0 x 10⁻⁵ Torr) was adopted.

RHEED was used to monitor the growth in-situ, and figure 2-4 show the patterns for the PbO, ZrO₂ and TiO₂, respectively.



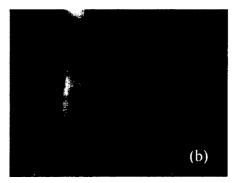
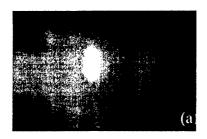


Figure.2 RHEED pattern of PbO grown on GaN (a) parallel to GaN ($11\overline{2}$ 0) direction (b) 15 degrees rotated from GaN ($11\overline{2}$ 0) direction



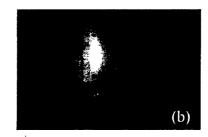
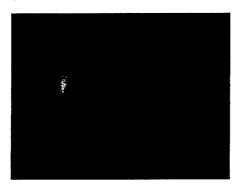


Figure.3 RHEED pattern of (a) ZrO₂low temperature buffer layer (b) high temperature layer both taken along (011) direction



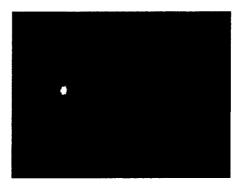


Figure.4 RHEED pattern of TiO_2 along GaN (11 $\overline{2}$ 0) (a) 10-nm-thick layer (b) 120-nm-thick layer.

As seen from figure 2, PbO shows a streaky pattern indicating 2 dimensional (2D) mode. In the case of ZrO₂, since the reactive species cannot acquire enough thermal energy at low growth temperatures, the ZrO₂ grown at low temperature exhibits a 3D RHEED

pattern, which turn into a 2-D streaky one upon annealing at 750°C and maintained during the high temperature growth. The situation is somewhat reverse for the TiO₂ growth: After 5 minutes growth the pattern of GaN diminished and replaced by the pattern of TiO₂. It is noticeable that the TiO₂ epitaxial on GaN started in a 2D mode. With the increase of the thickness the growth mode switched to 3D.

AFM was performed to study the surface morphology of epitaxially grown 10nm thick ZrO₂ and 20nm thick PbO layers, and the images are shown in figure 5. In both cases, the AFM images are indicative of smooth and uniform oxides. These oxides films are contiguous, and some features of the atomic step from the MOCVD GaN template remain observable.

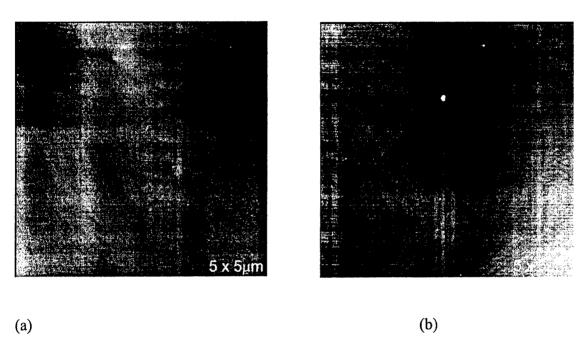
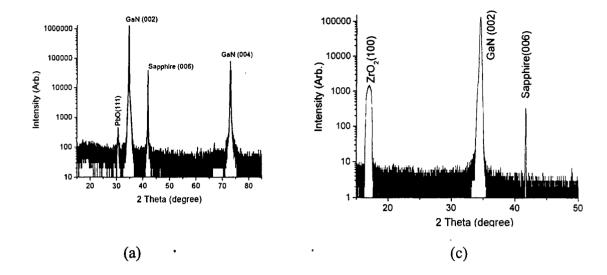


Figure.5 AFM images of (a) PbO on GaN (b) ZrO_2 on GaN. The image size is 5 μ m by 5 μ m and vertical scale is 5 μ m.

In order to study crystal structure and phase composition of the prepared oxide layers, XRD measurements have been performed and the patterns are shown in figure 6(a). For the 20-nm-thick PbO, only (111) reflections of orthorhombic PbO can be observed along with the GaN the sapphire substrate. The peaks of TiO₂ grown on GaN are consistent with (200) reflections of orthorhombic phase (see figure 6(b)). Figure 6(c) shows the 0-20 XRD pattern of a 30nm thick ZrO_2/GaN , which is indicative of (100)-oriented monoclinic ZrO_2 .



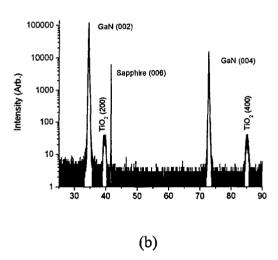


Figure.6 XRD theta-2theta pattern of (a) PbO(111)/GaN(0002) (b) ZrO₂(100)/GaN (0002) (c) TiO₂ (200)/GaN(0002)

Conclusions for ZrO₂

In summary, PbO, TiO₂ and ZrO₂ have been grown epitaxially on GaN/c-Al₂O₃ substrates by oxides MBE using H₂O₂ as a source of reactive oxygen. In situ RHEED was used to monitor the growth in-situ shows 2-D growth for PbO and ZrO₂, and a 2-D to 3-D transition for TiO₂. AFM was used to characterize the surface morphology of the thin PbO and ZrO₂ and indicates a contiguous feature. XRD pattern indicates that the growth orientation of these oxides are PbO [111]//GaN [0002], ZrO₂[100]//GaN [0002] and TiO₂[200]//GaN[0002].

Publications resulting from ONR support

- H. Lee, Y. -S. Kang, S-J Cho, Y.-I Alivov, N. Izyumskaya, and Hadis Morkoç, "Multifunctional perovskite oxides and their applications: a review", Critical Reviews in Solid State and Materials Sciences, by invitation only, submitted.
- Y.-I Alivov, N. Izyumskaya, F. Agra, M. Wu, and Hadis Morkoç, "Multifunctional oxides: ferroelectrics, ferromagnetics, multiferroics and high κ materials", Applied Physics Reviews, by invitation only about to be submitted.
- H. Lee, T. D. Kang, Y. S. Kang, S-J Cho, and H. Morkoç, "Visable-Deep UV Dielectric Functions and Electronic Band Structure of Lead Zirconate Titanate Thin Films", March 06 APS meeting, Baltimore Maryland

Hosun Lee, Youn Seon Kang, Sang-Jun Cho, Hadis Morkoç, and Tae Dong Kang, "Visible-ultraviolet spectroscopic ellipsometry of Lead Zirconate Titanate thin films", Appl. Phys. Lett. Appl. Phys. Lett. 86, 262902 (2005).

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V. Avrutin, N. Izyumskaya, Xing Gu, Ü. Özgür, B. Xiao, Tae Dong Kang, Hosun Lee, and H. Morkoç, "Growth of high quality Pb(Zr_xTi_{1-x})O₃ films by peroxide MBE and their optical and structural characteristics", Fall MRS 06 meeting

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